

weighed sample with 5 ml of Aquasol 2 in a scintillation counter. In the calculation of specific activity the weights were corrected for purity (see above).

Purification of gangliosides

GM₂ and GA₂ gangliosides were purified from the brain of a patient who died of Sandhoff's disease (14). The frozen brain was homogenized in 19 volumes (w/v) of chloroform-methanol 2:1 (v/v) and then solvent partitioned (15, 16). Most of the GM₂ ganglioside separated into the upper phase while the GA₂ ganglioside partitioned into the lower phase. The gangliosides were then treated separately in the following manner.

The ganglioside solution was repeatedly dried on a rotary evaporator and dissolved in chloroform-methanol 2:1 (v/v) to dissociate ganglioside from bound protein. The residues were dissolved in a minimum amount of chloroform and adsorbed on a silicic acid column (17). GA₂ ganglioside was eluted primarily with acetone-methanol while GM₂ ganglioside was found mostly in the methanol fraction. The gangliosides were then chromatographed on a silica gel H column (40 g activated at 120°C for 24 hr) with solvent system 1, and rechromatographed with solvent system 2. For final purification they were separated on silica gel H preparative thin-layer plates (500 μm) with solvent systems 1 and 2, and eluted with chloroform-methanol 2:1 containing 7% water (v/v). The purified gangliosides were dialyzed against water, lyophilized, and stored at 4°C.

Tritiation of gangliosides

The procedure follows a modification of the method of Radin et al. (1), treating GM₂ and GA₂ ganglioside separately in the same manner. Ganglioside (80 mg) was dissolved in 5 ml of freshly distilled tetrahydrofuran. Galactose oxidase (450 units) dissolved in 2 ml of potassium phosphate buffer (10 mM, pH 7) and horseradish peroxidase (2 mg) dissolved in 1 ml of buffer were added to the ganglioside, and the solution was stirred for 3 hr. Then a further 450 units of galactose oxidase in 2 ml of buffer and 1 mg of catalase in 1 ml of buffer were added and the solution was stirred overnight.

The oxidation reaction was stopped by the addition of 10 ml of chloroform-methanol 2:1 (v/v); the solution was mixed and centrifuged at 1,200 g for 10 min. For GM₂ ganglioside, the upper phase was removed and the lower phase was washed with 10 ml of ideal upper phase containing 0.1 N KCl (chloroform-methanol-water 3:48:47, v/v/v) followed by 10 ml of ideal upper phase containing water. The three upper phases were pooled. For GA₂ ganglio-

side, the lower phase was removed and the upper phase was washed with 10 ml of ideal lower phase (chloroform-methanol-water 86:14:1, v/v/v). The two lower phases were pooled. The ganglioside solution was dried and dissolved in a minimum amount of tetrahydrofuran. [³H]NaBH₄ (500 mCi, 8.2 Ci/mmol for GM₂; 100 mCi, 0.27 Ci/mmol for GA₂ ganglioside) was dissolved in 2 ml of 0.1 N NaOH and then was added to the ganglioside; the solution was stirred for 2 days at room temperature. Unlabeled NaBH₄ (50 mg) in 1 ml of 0.1 N NaOH was then added and the solution was stirred for 2 hr. Excess NaBH₄ was destroyed by adding approximately 2 ml of 10 N acetic acid dropwise in a fume hood, until the pH of the solution was acidic and no more hydrogen gas was produced. The mixture was dialyzed overnight against acetic acid-sodium acetate buffer (0.1 M, pH 5), then against water with several changes, and lyophilized. The tritiated gangliosides were purified from radioactive contamination by partition on a silica gel H column with solvent system 2 and by preparative TLC in solvent systems 1 and 2. They were lyophilized and stored at 4°C.

Identification of label

The radioactivity associated with the gangliosides was measured by TLC on silica gel G (250 μm) in solvent systems 1 and 2 with standards. Radioactive bands were visualized with a Panax TLC scanner and quantitated by scraping sections of lanes into a scintillation vial and counting in a scintillation counter with 5 ml of Aquasol 2 and 0.2 g of Cab-o-sil, to suspend the silica gel.

The determination of the distribution of label was based on the method of O'Brien et al. (3). For complete hydrolysis of ganglioside components, 20 nmol of tritiated GM₂ and GA₂ ganglioside were treated with 0.5 ml of 4 N HCl, gassed with nitrogen, and heated at 100°C for 4 hr. Partial hydrolysis of NeuNAc from GM₂ ganglioside was performed with 0.5 ml of 10 mM HCl and heated at 80°C for 1 hr to check for possible NeuNAc tritiation. The hydrolysates were dried, dissolved in chloroform-methanol-water 14:6:1 (v/v/v), and chromatographed on TLC in solvent system 3 with hydrolyzed standards. The hydrolysate lanes were scanned for radioactivity and quantitated by scraping and counting as before.

RESULTS AND DISCUSSION

The yield of purified unlabeled GM₂ and GA₂ ganglioside from frozen Sandhoff brain was about

0.4% and 0.1% (w/w), respectively. Both gangliosides ran as single bands on TLC. The R_f values in solvent system 1 are 0.45 for GM₂ ganglioside and 0.58 for GA₂ ganglioside, and in solvent system 2 they are 0.08 for GM₂ and 0.28 for GA₂.

The yields of tritiated gangliosides were about 25% for ³H-labeled GM₂ and 50% for ³H-labeled GA₂ ganglioside. Both preparations ran as single bands in solvent systems 1 and 2 with 96% of the radioactivity associated with GM₂ and 85% associated with GA₂ ganglioside. The rest of the counts were evenly distributed along the lanes with no other peaks showing on the radioscan. After strong acid hydrolysis of the gangliosides, 88% of the ³H-labeled GM₂ label and 91% of the ³H-labeled GA₂ label migrated on TLC with galactosamine. The other counts were evenly distributed over the lanes with no other peaks on the radioscan. Because the conditions of hydrolysis cause GalNAc to deacetylate, the study showed the gangliosides were labeled in the terminal GalNAc residue. Mild acid hydrolysis of ³H-labeled GM₂ ganglioside showed no label associated with NeuNAc.

The molar ratio of sphingosine to ³H-labeled GM₂ ganglioside was 1:1.0 and to ³H-labeled GA₂ ganglioside was 1:1.1. The molar ratio of bound NeuNAc to ³H-labeled GM₂ ganglioside was 1:0.9. In calculating specific activities, ³H-labeled GA₂ ganglioside was taken to be 100% pure and ³H-labeled GM₂ ganglioside to be 90% pure by weight.

It was more difficult to purify GM₂ than GA₂ ganglioside because the R_f values of GM₂ and GM₁ gangliosides are similar on TLC, while no major gangliosides migrate near GA₂. Also, the NeuNAc bond in GM₂ ganglioside is labile and NeuNAc may be hydrolyzed by normal purification procedure to produce GA₂ ganglioside. A sample of purified ³H-labeled GM₂ ganglioside left on a lyophilizer for 2 days showed significant breakdown into GA₂ ganglioside and had to be repurified by preparative TLC. Svennerholm (18) found that NeuNAc can be hydrolyzed from gangliosides during chromatography on a silicic acid column with polar organic solvents. O'Brien et al. (3) found that repeated freezing and thawing of ³H-labeled GM₂ ganglioside stored at -20°C in water or chloroform gave 0.5% ³H-labeled GA₂ contamination after 18 months.

We stored some small samples of tritiated gangliosides in chloroform-methanol 2:1 (v/v) and in citrate-phosphate buffer (30 mM, pH 4.1) at -20°C and 4°C for 2 months. After thawing the samples and running on TLC in solvent systems 1 and 2, only ³H-labeled GM₂ ganglioside stored at -20°C in chloroform-methanol 2:1 (v/v) showed contamina-

tion by radioscan. This peak corresponded to a small but significant amount of ³H-labeled GA₂ ganglioside. Our main supplies of both ³H-labeled GM₂ and ³H-labeled GA₂ ganglioside were stored as lyophilized powders at 4°C and neither showed contamination after 6 months of storage.

The final specific activity of the gangliosides is directly proportional to the amount of GalNAc oxidized by galactose oxidase and to the specific activity of [³H]NaBH₄ which, in excess, will reduce the oxidized GalNAc residues 1:4 on a molar basis (19). Therefore we can calculate the percent ganglioside oxidized, which is equal to the percent ganglioside reduced by $(4 \times \text{specific activity of } ^3\text{H-labeled ganglioside} / \text{specific activity of } [^3\text{H}]\text{NaBH}_4) \times 100$.

With the addition of catalase and peroxidase into the galactose oxidase step, the specific activity of our GM₂ ganglioside was 4.62×10^8 dpm/ μ mol (2.08×10^{-4} Ci/ μ mol). This corresponds to oxidation of 10.2% of the GalNAc residues. In one preparation we utilized the same procedure to tritiate GM₂ ganglioside (with 500 mCi of [³H]NaBH₄, 11.1 Ci/mmol) using horseradish peroxidase, but omitting catalase. We obtained a lower specific activity of 7.20×10^7 dpm/ μ mol (3.24×10^{-5} Ci/ μ mol) which can be calculated as 1.2% oxidation. Comparing our results with those of others who used a similar technique, Hechtman (8) adding only catalase, obtained 0.04% oxidation and a final specific activity of 2.50×10^6 dpm/ μ mol GM₂. Without using catalase or peroxidase, the highest specific activity attained for GM₂ ganglioside was 6×10^7 dpm/ μ mol, giving 5.7% oxidation (3). The specific activity of our GA₂ ganglioside, using catalase and peroxidase, is 5.54×10^7 dpm/ μ mol (2.50×10^{-5} Ci/ μ mol), indicating 37.0% oxidation and confirming that GA₂ is better than GM₂ ganglioside as a substrate for this reaction (2, 3). We have described a good procedure for purifying GM₂ and GA₂ ganglioside and a simple method of increasing the specific activity when lipids are tritiated using the galactose oxidase-[³H]NaBH₄ procedure. ■

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